THE SYNTHESIS OF ANOMERIC METHYL 4-O-L-MYCAROSYL-β-D-MYCAMINOSIDES

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The anomeric forms of methyl 4-0-L-mycarosyl- β -D-mycaminoside (VII) were synthesized by the condensation of 3,4-0-carbonyl-L-mycarosyl chloride and methyl N-carboethoxy-des-N-methyl-2-0-p-phenylazobenzoyl- β -D-mycaminoside (III) and following removal of the acyl group and N-methylation.

Recently, we have carried out the glycosidation of L-mycarose, a neutral branched-chain sugar found in macrolide antibiotics, with the primary hydroxyl group of D-glucosamine Now, we wish to report an example of the glycosidation of L-mycarose with the secondary hydroxyl group: the synthesis of the anomeric forms of methyl 4-0-L-mycarosyl- β -D-mycaminoside (VII). The α -anomer (VIIa) is the disaccharide constituting some kind of the 16-membered macroride antibiotics 2 .

 β -D-Mycaminose hydrochloride monohydrate³⁾ was treated with acetyl bromide⁴⁾ to give a bromide (I) in the form of hydrobromide, which was methanolyzed in the presence of, at first, silver oxide and then sodium methoxide to give methyl β -D-mycaminoside (II)⁵⁾. II was treated with chloroethylformate in the presence of sodium carbonate⁶⁾, hydrolyzed with sodium ethoxide, and then selectively azoylated with azoyl chloride in pyridine to give methyl N-carboethoxy-des-N-methyl-2-0-p-phenylazobenzoyl- β -D-mycaminoside (III); the C-2 substitution was checked by PMR at 100 MHz in C₆D₆.

III was condensed with 3,4-0-carbonyl-L-mycarosyl chloride 1 in nitromethane in the presence of mercuric cyanide to afford methyl 4-0-(3,4-0-carbonyl- α -L-mycarosyl)-N-carboethoxy-des-N-methyl-2-0-p-phenylazobenzoyl- β -D-mycaminoside (IVa) as the main product (11%), as well as the β -anomer (IVb: 7%). IVa was successively ethanolyzed, hydrolyzed with hot barium hydroxide, and then N-methylated reductively (formaline/palladium/ethanol) to give methyl 4-0- α -L-mycarosyl- β -D-mycaminoside (VIIa).

The minor product, TVb was treated as just above to furnish the $\beta\mbox{-anomer}$ of the titled glycoside (VIIb).

Thus, the ratio of the formation of the both anomers were 3 : 2; the predominant formation of α -glycoside was again observed $^{1)}$.

References.

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$$\begin{array}{c} \begin{array}{c} \begin{array}{c} CH_3 \\ OOH \end{array} \\ OOH \\$$

PHYSICAL DATA

Compounds	mp (°C)	$\left[\alpha\right]_{D}^{25}$ (%, solvent)
I	72 76	+100° (0.7, chloroform)
II	120 123	-35° (1.4, water)
III	134 136	-57° (1.3, chloroform)
IVa	97 99	$+1.6^{\circ}(1.0, \text{chloroform})$
IVb	176 178	$+105^{\circ}$ (1.0, chloroform)
Va	94 96	-84° (1.0, chloroform)
Vb	137 139°	$+17^{\circ}$ (1.0, chloroform)
VIa	glass	-78° (0.7, methanol)
VIb	195 197°	$+8^{\circ}$ (0.7, methanol)
VIIa	glass	-93° (1.1, ethanol)
VIIb	glass	+14° (0.9, ethanol)

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